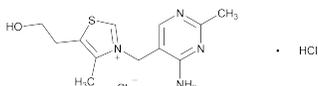


the peak responses obtained from the *Assay preparation* and the *Standard preparation*, respectively.

## Thiamine Hydrochloride



$C_{12}H_{17}ClN_4OS \cdot HCl$  337.27  
Thiazolium, 3-[(4-amino-2-methyl-5-pyrimidinyl)methyl]-5-(2-hydroxyethyl)-4-methyl-, chloride, monohydrochloride; Thiamine monohydrochloride [67-03-8].

### DEFINITION

Thiamine Hydrochloride contains NLT 98.0% and NMT 102.0% of thiamine hydrochloride ( $C_{12}H_{17}ClN_4OS \cdot HCl$ ), calculated on the anhydrous basis.

### IDENTIFICATION

- A. INFRARED ABSORPTION** (197K)  
**Analysis:** Dry specimens at 105° for 2 h.  
**Acceptance criteria:** Meets the requirements
- B. IDENTIFICATION TESTS—GENERAL, Chloride** (191): A 20-mg/mL solution meets the requirements.

### ASSAY

#### PROCEDURE

**Solution A:** 0.005 M sodium 1-octanesulfonate in dilute glacial acetic acid (1 in 100)

**Solution B:** Methanol and acetonitrile (3:2)

**Mobile phase:** *Solution B* and *Solution A* (40:60)

**Internal standard solution:** 2% (v/v) of methylbenzoate in methanol

**Standard solution:** Prepare a 1-mg/mL solution of USP Thiamine Hydrochloride RS in *Mobile phase*. Transfer 20.0 mL of this solution and 5.0 mL of *Internal standard solution* to a 50-mL volumetric flask, and dilute with *Mobile phase* to volume. The *Standard solution* contains 400 µg/mL of thiamine hydrochloride.

**Sample solution:** Prepare a 2-mg/mL of Thiamine Hydrochloride in *Mobile phase*. Transfer 10.0 mL of this solution and 5.0 mL of *Internal standard solution* to a 50-mL volumetric flask, and dilute with *Mobile phase* to volume.

#### Chromatographic system

(See *Chromatography* (621), *System Suitability*.)

**Mode:** LC

**Detector:** UV 254 nm

**Column:** 4-mm × 30-cm; packing L1

**Flow rate:** 1 mL/min

[NOTE—The flow rate may be adjusted as needed to obtain a retention time of about 12 min for thiamine hydrochloride.]

**Injection size:** 10 µL

#### System suitability

**Sample:** *Standard solution*

#### Suitability requirements

**Resolution:** NLT 4.0 between the thiamine and methylbenzoate peaks

**Tailing factor:** NMT 2.0 for the thiamine peak

**Column efficiency:** NLT 1500 theoretical plates for the thiamine peak

**Relative standard deviation:** NMT 2.0% for the ratios of thiamine peak areas to the internal standard peak area

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of thiamine hydrochloride ( $C_{12}H_{17}ClN_4OS \cdot HCl$ ) in the portion of Thiamine Hydrochloride taken:

$$\text{Result} = (R_U/R_S) \times (C_S/C_U) \times 100$$

$R_U$  = internal standard ratio (peak area of thiamine/peak area of the internal standard) from the *Sample solution*

$R_S$  = internal standard ratio (peak area of thiamine/peak area of the internal standard) from the *Standard solution*

$C_S$  = concentration of USP Thiamine Hydrochloride RS in the *Standard solution* (mg/mL)

$C_U$  = concentration of Thiamine Hydrochloride in the *Sample solution* (mg/mL)

**Acceptance criteria:** 98.0%–102.0% on the anhydrous basis

### IMPURITIES

- RESIDUE ON IGNITION** (281): NMT 0.2%

#### LIMIT OF NITRATE

**Sample solution:** 20 mg/mL of Thiamine Hydrochloride

**Analysis:** To 2 mL of the *Sample solution* add 2 mL of sulfuric acid. Cool, and superimpose 2 mL of ferrous sulfate TS.

**Acceptance criteria:** No brown ring is produced at the junction of the two layers.

#### RELATED COMPOUNDS

**Solution A, Solution B, and Mobile phase:** Proceed as directed in the *Assay*.

**Sample solution:** 1.0 mg/mL of Thiamine Hydrochloride in *Mobile phase*

#### Chromatographic system

(See *Chromatography* (621), *System Suitability*.)

**Mode:** LC

**Detector:** UV 254 nm

**Column:** 4.0-mm × 15-cm; packing L1

**Flow rate:** 0.75 mL/min

**Injection size:** 10 µL

#### Analysis

**Sample:** *Sample solution*

Allow the *Sample solution* to elute for NLT three times the retention time of the main peak.

Calculate the percentage of total secondary peaks in the portion of Thiamine Hydrochloride taken:

$$\text{Result} = (r_U/r_T) \times 100$$

$r_U$  = sum of the areas of all the peaks, except that of the thiamine peak

$r_T$  = sum of the areas of all the peaks

**Acceptance criteria:** NMT 1.0%

### SPECIFIC TESTS

#### PH

**Sample solution:** 10-mg/mL solution

**Acceptance criteria:** 2.7–3.4

#### WATER DETERMINATION, Method I

(921): NMT 5.0%

#### ABSORBANCE OF SOLUTION

**Sample solution:** 100 mg/mL in water. Filter through a fine-porosity, sintered-glass funnel.

**Blank:** Water

#### Instrumental conditions

(See *Spectrophotometry and Light-Scattering* (851).)

**Mode:** UV-Vis

**Analytical wavelength:** 400 nm

**Cell:** 1 cm

#### Analysis

**Samples:** *Sample solution* and *Blank*

Determine the absorbance of the *Sample solution* against that of the *Blank*.

**Acceptance criteria:** NMT 0.025

### ADDITIONAL REQUIREMENTS

- PACKAGING AND STORAGE:** Preserve in tight, light-resistant containers.

- **USP REFERENCE STANDARDS** (11)  
USP Thiamine Hydrochloride RS

## Thiamine Hydrochloride Injection

» Thiamine Hydrochloride Injection is a sterile solution of Thiamine Hydrochloride in Water for Injection. It contains not less than 90.0 percent and not more than 110.0 percent of the labeled amount of thiamine hydrochloride ( $C_{12}H_{17}ClN_4OS \cdot HCl$ ).

**Packaging and storage**—Preserve in single-dose or multiple-dose containers, preferably of Type I glass, protected from light.

### USP Reference standards (11)—

USP Endotoxin RS

USP Thiamine Hydrochloride RS

### Identification—

**A:** It yields a white precipitate with mercuric chloride TS, and a red-brown precipitate with iodine TS. It also yields a precipitate with mercuric-potassium iodide TS, and with trinitrophenol TS.

**B:** Dilute a portion of Injection with water to a concentration of about 10 mg of thiamine hydrochloride per mL. To 0.5 mL of this solution add 5 mL of 0.5 N sodium hydroxide, then add 0.5 mL of potassium ferricyanide TS and 5 mL of isobutyl alcohol, shake the mixture vigorously for 2 minutes, and allow the liquid layers to separate: when illuminated from above by a vertical beam of UV light and viewed at a right angle to this beam, the air-liquid meniscus shows a vivid blue fluorescence, which disappears when the mixture is slightly acidified, but reappears when it is again made alkaline.

**C:** It responds to the tests for *Chloride* (191).

**Bacterial endotoxins** (85)—It contains not more than 3.5 USP Endotoxin Units per mg of thiamine hydrochloride.

**pH** (791): between 2.5 and 4.5.

**Other requirements**—It meets the requirements under *Injections* (1).

### Assay—

*Mobile phase, Internal standard solution, Standard preparation, and Chromatographic system*—Prepare as directed in the *Assay* under *Thiamine Hydrochloride Oral Solution*.

*Assay preparation*—Quantitatively dilute an accurately measured volume of Injection with *Mobile phase* to obtain a solution containing about 500 µg of thiamine hydrochloride per mL. Pipet 10 mL of the resulting solution and 10 mL of *Internal standard solution* into a 100-mL volumetric flask, dilute with *Mobile phase* to volume, and mix.

*Procedure*—Proceed as directed for *Procedure* in the *Assay* under *Thiamine Hydrochloride Oral Solution*. Calculate the quantity, in mg, of thiamine hydrochloride ( $C_{12}H_{17}ClN_4OS \cdot HCl$ ) in each mL of the Injection taken by the formula:

$$C(L/D)(R_u / R_s)$$

in which C is the concentration, in mg per mL, of USP Thiamine Hydrochloride RS in the *Standard preparation*; L is the labeled quantity, in mg per mL, of thiamine hydrochloride in the Injection; D is the concentration, in mg per mL, of thiamine hydrochloride in the *Assay preparation* on the basis of the labeled quantity and the extent of dilution; and  $R_u$  and  $R_s$  are the ratios of the peak responses of thiamine to methylparaben obtained from the *Assay preparation* and the *Standard preparation*, respectively.

## Thiamine Hydrochloride Oral Solution

### DEFINITION

Thiamine Hydrochloride Oral Solution contains NLT 95.0% and NMT 135.0% of the labeled quantity of thiamine hydrochloride ( $C_{12}H_{17}ClN_4OS \cdot HCl$ ).

### IDENTIFICATION

#### • A.

**Sample solution:** Dilute a portion of Oral Solution with water to a concentration of 10 mg/mL of thiamine hydrochloride.

**Analysis:** To 0.5 mL of the *Sample solution* add 5 mL of 0.5 N sodium hydroxide, then add 0.5 mL of potassium ferricyanide TS and 5 mL of isobutyl alcohol. Shake the mixture vigorously for 2 min, and allow the liquid layers to separate. Illuminate from above by a vertical beam of UV light, and observe the air-liquid meniscus at a right angle to this beam.

**Acceptance criteria:** The air-liquid meniscus shows a vivid blue fluorescence, which disappears when the mixture is slightly acidified, but reappears when it is again made alkaline.

### ASSAY

#### • PROCEDURE

**Mobile phase:** Methanol and 0.04 M aqueous monobasic potassium phosphate (45:55)

**Internal standard solution:** 100 µg/mL of methylparaben in *Mobile phase*

**Standard stock solution:** 500 µg/mL of USP Thiamine Hydrochloride RS in *Mobile phase*

**Standard solution:** Dilute a mixture of equal volumes of the *Standard stock solution* and *Internal standard solution* with *Mobile phase* to obtain a concentration of USP Thiamine Hydrochloride RS of about 50 µg/mL.

**Sample stock solution:** Equivalent to 500 µg/mL of thiamine hydrochloride in *Mobile phase* from an accurately measured volume of Oral Solution

**Sample solution:** Dilute a mixture of equal volumes of the *Sample stock solution* and *Internal standard solution* with *Mobile phase* to obtain a concentration of thiamine hydrochloride of about 50 µg/mL.

#### Chromatographic system

(See *Chromatography* (621), *System Suitability*.)

**Mode:** LC

**Detector:** UV 254 nm

**Column:** 3.9-mm × 30-cm; packing L1

**Flow rate:** 1 mL/min

**Injection size:** 25 µL

#### System suitability

**Sample:** *Standard solution*

[NOTE—The relative retention times for thiamine and methylparaben are about 0.35 and 1.0, respectively.]

#### Suitability requirements

**Resolution:** NLT 6.0 between thiamine and methylparaben

**Relative standard deviation:** NMT 2.0%

#### Analysis

**Samples:** *Standard solution* and *Sample solution*

Calculate the percentage of the labeled amount of thiamine hydrochloride ( $C_{12}H_{17}ClN_4OS \cdot HCl$ ) in the portion of Oral Solution taken:

$$\text{Result} = (R_u/R_s) \times (C_s/C_u) \times 100$$

$R_u$  = peak area ratio of thiamine to methylparaben from the *Sample solution*

$R_s$  = peak area ratio of thiamine to methylparaben from the *Standard solution*

$C_s$  = concentration of USP Thiamine Hydrochloride RS in the *Standard stock solution* (µg/mL)

$C_u$  = nominal concentration of thiamine hydrochloride in the *Sample stock solution* (µg/mL)