

acid, stopper tightly at once, shake, and allow to stand in a dark place for 60 minutes. Add carefully 50 mL of water and 15 mL of potassium iodide TS, stopper tightly at once, shake well, and titrate the liberated iodine with 0.1 mol/L sodium thiosulfate VS (indicator: 1 mL of starch TS). Perform a blank determination.

Each mL of 0.05 mol/L bromine VS
= 15.157 mg of $C_{13}H_{12}Cl_2O_4$

Containers and storage Containers—Well-closed containers.

Etacrylic Acid Tablets

エタクリン酸錠

Etacrylic Acid Tablets contain not less than 90% and not more than 110% of the labeled amount of etacrylic acid ($C_{13}H_{12}Cl_2O_4$: 303.14).

Method of preparation Prepare as directed under Tablets, with Etacrylic Acid.

Identification (1) Weigh a quantity of powdered Etacrylic Acid Tablets, equivalent to 0.3 g of Etacrylic Acid according to the labeled amount, add 25 mL of 0.1 mol/L hydrochloric acid TS, and extract with 50 mL of Dichloromethane. Filter the dichloromethane extract, and evaporate the filtrate on a water bath to dryness. Proceed with the residue as directed in the Identification (1), (2) and (4) under Etacrylic Acid.

(2) Prepare a solution of the residue obtained in (1), equivalent to a solution of Etacrylic Acid in methanol (1 in 20,000), and determine the absorption spectrum as directed under the Ultraviolet-visible Spectrophotometry: it exhibits a maximum between 268 nm and 272 nm.

Dissolution test Perform the test with 1 tablet of Etacrylic Acid Tablets at 50 revolutions per minute according to Method 2 under the Dissolution Test, using 900 mL of water as the test solution. Take 20 mL or more of the dissolved solution 45 minutes after starting the test, and filter through a membrane filter with pore size of not more than 0.8 μm . Discard the first 10 mL of the filtrate, and use the subsequent as the sample solution. Separately, weigh accurately about 0.055 g of etacrylic acid for assay, previously dried at 60°C for 2 hours, dissolve in 10 mL of methanol, and add water to make exactly 100 mL. Pipet 5 mL of this solution, add water to make exactly 100 mL, and use this solution as the standard solution. Determine the absorbances, A_T and A_S , of the sample solution and the standard solution at 277 nm as directed under the Ultraviolet-visible Spectrophotometry.

The dissolution rate of Etacrylic Acid Tablets in 45 minutes is not less than 70%.

Dissolution rate (%) with respect to
the labeled amount of etacrylic acid ($C_{13}H_{12}Cl_2O_4$)

$$= W_s \times \frac{A_T}{A_S} \times \frac{V'}{V} \times \frac{1}{C} \times 900$$

W_s : Amount (mg) of etacrylic acid for assay.

C : Labeled amount (mg) of etacrylic acid ($C_{13}H_{12}Cl_2O_4$) in 1 tablet.

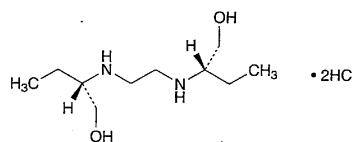
Assay Weigh accurately and powder not less than 20 Etacrylic Acid Tablets. Weigh accurately a portion of the powder, equivalent to about 0.1 g of etacrylic acid ($C_{13}H_{12}Cl_2O_4$), add 25 mL of 0.1 mol/L hydrochloric acid TS, and extract with three 30-mL portions of dichloromethane. Filter the dichloromethane extracts through a pledget of absorbent cotton into an iodine bottle. Wash the pledget of absorbent cotton with a small amount of dichloromethane, and combine the washing with the extracts. Evaporate this solution on a water bath to dryness in a current of air, to the residue add 20 mL of acetic acid (100), and proceed as directed in the Assay under Etacrylic Acid.

Each mL of 0.05 mol/L bromine VS
= 15.157 mg of $C_{13}H_{12}Cl_2O_4$

Containers and storage Containers—Well-closed containers.

Ethambutol Hydrochloride

塩酸エタンブトール



$C_{10}H_{24}N_2O_2 \cdot 2HCl$: 277.23

N,N'-Ethylenebis[2*S*]-2-aminobutanol] dihydrochloride
[1070-11-7]

Ethambutol Hydrochloride, when dried, contains not less than 98.5% of $C_{10}H_{24}N_2O_2 \cdot 2HCl$.

Description Ethambutol Hydrochloride occurs as white crystals or crystalline powder. It is odorless, and has a bitter taste.

It is very soluble in water, soluble in methanol and in ethanol (95), and practically insoluble in diethyl ether.

The pH of a solution of Ethambutol Hydrochloride (1 in 20) is between 3.4 and 4.0.

Identification (1) To 10 mL of a solution of Ethambutol Hydrochloride (1 in 100) add 0.5 mL of copper (II) sulfate TS and 2 mL of sodium hydroxide TS: a deep blue color is produced.

(2) Dissolve 0.1 g of Ethambutol Hydrochloride in 40 mL of water, add 20 mL of 2,4,6-trinitrophenol TS, and allow to stand for 1 hour. Collect the precipitate, wash with 50 mL of water, and dry at 105°C for 2 hours: the precipitate melts between 193°C and 197°C.

(3) A solution of Ethambutol Hydrochloride (1 in 30) responds to the Qualitative Tests for chloride.

Optical rotation $[\alpha]_D^{20}$: +5.5 – +6.1° (after drying, 5 g, water, 50 mL, 200 mm).

Melting point 200 – 204°C

Purity (1) Clarity and color of solution—Dissolve 1.0 g of Ethambutol Hydrochloride in 10 mL of water: the solution is clear and colorless.

(2) Heavy metals—Proceed with 2.0 g Ethambutol