## Croconazole Hydrochloride

塩酸クロコナゾール

C<sub>18</sub>H<sub>15</sub>ClN<sub>2</sub>O.HCl: 347.24 1-{1-[2-(3-Chlorobenzyloxy)phenyl]vinyl}-1*H*-imidazole monohydrochloride [77174-66-4]

Croconazole Hydrochloride, when dried, contains not less than 98.5% of  $C_{18}H_{15}CIN_2O.HCl.$ 

**Description** Croconazole Hydrochloride occurs as white to pale yellowish white crystals or crystalline powder.

It is very soluble in water, freely soluble in methanol, in ethanol (95) and in acetic acid (100), and practically insoluble in diethyl ether.

**Identification** (1) Determine the absorption spectrum of a solution of Croconazole Hydrochloride in methanol (1 in 20,000) as directed under the Ultraviolet-visible Spectrophotometry, and compare the spectrum with the Reference Spectrum: both spectra exhibit similar intensities of absorption at the same wavelengths.

- (2) Determine the infrared absorption spectrum of Croconazole Hydrochloride, previously dried, as directed in the potassium chloride disk method under the Infrared Spectrophotometry, and compare the spectrum with the Reference Spectrum: both spectra exhibit similar intensities of absorption at the same wave numbers.
- (3) Dissolve 0.05 g of Croconazole Hydrochloride in 10 mL of water, add 2 mL of sodium hydroxide TS and 20 mL of diethyl ether, and shake. Wash the separated aqueous layer with two 10-mL portions of diethyl ether, and acidify the solution with 2 mL of dilute nitric acid: the solution responds to the Qualitative Tests for chloride.

## Melting point 148 – 153°C

- **Purity** (1) Heavy metals—Proceed with 1.0 g of Croconazole Hydrochloride according to Method 4, and perform the test. Prepare the control solution with 1.0 mL of Standard Lead Solution (not more than 10 ppm).
- (2) Related substances—Dissolve 0.050 g of Croconazole Hydrochloride in 10 mL of methanol, and use this solution as the sample solution. Pipet 1 mL of the sample solution, add methanol to make exactly 100 mL, and use this solution as the standard solution. Perform the test with these solutions as directed under the Thin-layer Chromatography. Spot  $10 \,\mu\text{L}$  each of the sample solution and the standard solution on a plate of silica gel with fluorescent indicator for thin-layer chromatography. Develop the plate with a mixture of ethyl acetate, hexane, methanol and ammonia solution (28) (30:15:5:1) to a distance of about 10 cm, and airdry the plate. Examine under ultraviolet light (main wavelength: 254 nm): the spots other than the principal spot

and other than the spot of the starting point from the sample solution are not more intense than the spot from the standard solution.

Loss on drying Not more than 0.5% (1 g, 60°C, 4 hours).

Residue on ignition Not more than 0.10% (1 g).

Assay Weigh accurately about 0.6 g of Croconazole Hydrochloride, previously dried, dissolve in 10 mL of acetic acid (100), add 40 mL of acetic anhydride, and titrate with 0.1 mol/L perchloric acid VS [indicator: 1 to 2 drops of a solution of malachite green oxalate in acetic acid (100) (1 in 100)] until the color of the solution changes from blue-green through green to yellow-green. Perform a blank determination, and make any necessary correction.

Each mL of 0.1 mol/L perchloric acid VS = 34.724 mg of C<sub>18</sub>H<sub>15</sub>ClN<sub>2</sub>O.HCl

**Containers and storage** Containers—Tight containers. Storage—Light-resistant.

## Cyanamide

シアナミド

H<sub>2</sub>N-CN

CH<sub>2</sub>N<sub>2</sub>: 42.04 Aminonitrile [420-04-2]

Cyanamide contains not less than 97.0% of  $CH_2N_2$ , calculated on the anhydrous basis.

**Description** Cyanamide occurs as white crystals or crystalline powder. It has a faint, characteristic odor.

It is very soluble in water, in methanol and in ethanol (95), and freely soluble in diethyl ether.

The pH of a solution of Cyanamide (1 in 100) is between 5.0 and 6.5.

It is hygroscopic.

Melting point: about 46°C

**Identification** (1) To 5 mL of a solution of Cyanamide (1 in 100) add 2 mL of a solution of sodium pentacyanoammine ferroate (II) *n*-hydrate (1 in 100): a red-purple color develops.

- (2) To 1 mL of a solution of Cyanamide (1 in 100) add 1 mL of potassium 1,2-naphthoquinone-4-sulfonate TS and 0.2 mL of sodium hydroxide TS: a deep red color develops.
- (3) To 5 mL of a solution of Cyanamide (1 in 100) add 1 mL of ammonia TS and 1 mL of silver nitrate TS: a yellow precipitate is formed.
- **Purity** (1) Clarity and color of solution—Dissolve 1.0 g of Cyanamide in 10 mL of water: the solution is clear and colorless.
- (2) Sulfate—Perform the test with 0.5 g of Cyanamide. Prepare the control solution with 0.40 mL of 0.005 mol/L sulfuric acid VS (not more than 0.038%).
- (3) Heavy metals—Proceed with 2.0 g of Cyanamide according to Method 1, and perform the test. Prepare the control solution with 2.0 mL of Standard Lead Solution (not more than 10 ppm).