ASSAY

Transfer 1.000 g into a ground-glass-stoppered flask containing 150 mL of *water R* and stir for 1 h. Add 0.1 mL of *dilute acetic acid R* and titrate with $0.1\,M$ sodium thiosulfate using starch solution R as indicator.

1 mL of 0.1 M sodium thiosulfate is equivalent to 12.69 mg of available iodine.

STORAGE

Protected from light.

07/2010:2416

PRAMIPEXOLE DIHYDROCHLORIDE MONOHYDRATE

Pramipexoli dihydrochloridum monohydricum

C₁₀H₁₉Cl₂N₃S,H₂O [191217-81-9] $M_{\rm r}$ 302.3

DEFINITION

(6*S*)-6*N*-Propyl-4,5,6,7-tetrahydro-1,3-benzothiazole-2,6-diamine dihydrochloride monohydrate.

Content: 99.0 per cent to 101.0 per cent (anhydrous substance).

CHARACTERS

Appearance: white or almost white, crystalline powder. Solubility: freely soluble in water, soluble in methanol, sparingly soluble to slightly soluble in ethanol (96 per cent) and practically insoluble in methylene chloride.

IDENTIFICATION

Carry out either tests B, C, D or tests A, B, D.

A. Specific optical rotation (2.2.7): -67.0 to -69.5 (anhydrous substance).

Dissolve 0.250 g in $\it methanol~R$ and dilute to 25.0 mL with the same solvent.

- B. Infrared absorption spectrophotometry (2.2.24). Comparison: pramipexole dihydrochloride monohydrate CRS.
- C. Enantiomeric purity (see Tests).
- D. It gives reaction (a) of chlorides (2.3.1).

TESTS

Appearance of solution. The solution is clear (2.2.1) and not more intensely coloured than reference solution Y_6 (2.2.2, Method II).

Dissolve 0.1 g in $water\ R$ and dilute to 10 mL with the same solvent.

pH (2.2.3): 2.8 to 3.4.

Dissolve $0.4~{\rm g}$ in carbon dioxide-free water R and dilute to $20~{\rm mL}$ with the same solvent.

Related substances. Liquid chromatography (2.2.29).

Buffer solution. Dissolve 5 g of sodium octanesulfonate monohydrate R and 9.1 g of potassium dihydrogen phosphate R in 900 mL of water R. Adjust to pH 3.0 with phosphoric acid R and dilute to 1000 mL with water R.

Solvent mixture: acetonitrile R, buffer solution (200:800 V/V). Test solution. Dissolve 75 mg of the substance to be examined in the solvent mixture and dilute to 50.0 mL with the solvent mixture.

Reference solution (a). Dilute 1.0 mL of the test solution to 100.0 mL with the solvent mixture. Dilute 1.0 mL of this solution to 10.0 mL with the solvent mixture.

Reference solution (b). Dissolve 7.5 mg of *pramipexole for system suitability CRS* (containing impurities A, B and C) in 5.0 mL of the solvent mixture.

Column

- size: l = 0.125 m, $\emptyset = 4.6$ mm;
- stationary phase: end-capped octadecylsilyl silica gel for chromatography R (5 µm);
- temperature: 40 °C.

Mobile phase:

- mobile phase A: buffer solution;
- mobile phase B: acetonitrile R, buffer solution (500:500 V/V);

Time (min)	Mobile phase A (per cent <i>V/V</i>)	Mobile phase B (per cent <i>V/V</i>)
0 - 15	$60 \rightarrow 20$	$40 \rightarrow 80$

Flow rate: 1.5 mL/min.

Detection: spectrophotometer at 264 nm.

Injection: 5 µL.

Identification of impurities: use the chromatogram supplied with *pramipexole for system suitability CRS* and the chromatogram obtained with reference solution (b) to identify the peaks due to impurities A, B and C.

Relative retention with reference to pramipexole (retention time = about 6 min): impurity A = about 0.7; impurity B = about 1.5; impurity C = about 1.7.

System suitability: reference solution (b):

 resolution: minimum 6.0 between the peaks due to impurity A and pramipexole.

Limits:

- impurities A, B, C: for each impurity, not more than 1.5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.15 per cent);
- unspecified impurities: for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (a) (0.10 per cent);
- total: not more than 5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.5 per cent);
- disregard limit: 0.5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.05 per cent).

Enantiomeric purity. Liquid chromatography (2.2.29).

Test solution. Dissolve 6 mg of the substance to be examined in 5 mL of *anhydrous ethanol* R and dilute to 20.0 mL with the mobile phase.

Reference solution (a). Dissolve 2 mg of pramipexole impurity D CRS in the mobile phase and dilute to 10 mL with the mobile phase. To 1 mL of this solution add 1 mL of the test solution and dilute to 20 mL with the mobile phase.

Reference solution (b). Dilute 1.0 mL of the test solution to 20.0 mL with the mobile phase. Dilute 1.0 mL of this solution to 10.0 mL with the mobile phase.

Column:

- size: l = 0.25 m, $\emptyset = 4.6$ mm;

stationary phase: silica gel OD for chiral separations R.

Mobile phase: diethylamine R, anhydrous ethanol R, hexane R (1:150:850 V/V/V).

Flow rate: 1.5 mL/min.

Detection: spectrophotometer at 254 nm.

Injection: 75 µL.

Run time: 1.5 times the retention time of pramipexole.

Relative retention with reference to pramipexole (retention time = about 11 min): impurity D = about 0.5.

System suitability:

- resolution: minimum 5 between the peaks due to impurity D and pramipexole in the chromatogram obtained with reference solution (a);
- symmetry factor: maximum 2.4 for the peak due to pramipexole in the chromatogram obtained with reference solution (b).

Limit:

 impurity D: not more than the area of the principal peak in the chromatogram obtained with reference solution (b) (0.5 per cent).

Heavy metals (2.4.8): maximum 10 ppm.

Solvent: water R.

0.500 g complies with test H. Prepare the reference solution using 0.5 ml of *lead standard solution* (10 ppm Pb) R.

Water (2.5.12): 4.5 per cent to 6.5 per cent, determined on 0.500 g.

Sulfated ash (2.4.14): maximum 0.1 per cent, determined on 1.0 g.

ASSAY

Dissolve 0.120 g in 150 mL of *water R*. Add 10 mL of *3 M nitric acid* and titrate with 0.1 M silver nitrate, determining the end-point potentiometrically (2.2.20).

1 mL of 0.1 M silver nitrate is equivalent to 14.213 mg of $\rm C_{10}H_{19}Cl_2N_3S$.

IMPURITIES

Specified impurities: A, B, C, D.

Other detectable impurities (the following substances would, if present at a sufficient level, be detected by one or other of the tests in the monograph. They are limited by the general acceptance criterion for other/unspecified impurities and/or by the general monograph Substances for pharmaceutical use (2034). It is therefore not necessary to identify these impurities for demonstration of compliance. See also 5.10. Control of impurities in substances for pharmaceutical use): E.

$$H_2N$$
 N N N

A. (6S)-4,5,6,7-tetrahydro-1,3-benzothiazole-2,6-diamine,

B. (6S)-N,N'-dipropyl-4,5,6,7-tetrahydro-1,3-benzothiazole-2,6-diamine,

$$H_2N$$
 S
 H_3C
 H_3C
 H_3C
 H_3C
 H_3C
 H_3C
 H_3C

C. mixture of diastereoisomers of (6S)-6-N-[3-[(6S)-2-amino-4,5, 6,7-tetrahydro-1,3-benzothiazol-6-yl]-1-ethyl-2-methylpropyl]-4,5,6,7-tetrahydro-1,3-benzothiazole-2,6-diamine,

D. (6R)-6-N-propyl-4,5,6,7-tetrahydro-1,3-benzothiazole-2,6-diamine.

E. *N*-[(6*S*)-2-amino-4,5,6,7-tetrahydro-1,3-benzothiazol-6-yl]propanamide.

01/2010:2059

PRAVASTATIN SODIUM

Pravastatinum natricum

 $C_{23}H_{35}NaO_7$ [81131-70-6]

 $M_{\rm r}$ 446.5

DEFINITION

Sodium (3R,5R)-3,5-dihydroxy-7-[(1S,2S,6S,8S,8aR)-6-hydroxy-2-methyl-8-[[(2S)-2-methylbutanoyl]oxy]-1,2,6,7,8,8a-hexahydronaphthalen-1-yl]heptanoate.

Content: 97.0 per cent to 102.0 per cent (anhydrous substance).

CHARACTERS

Appearance: white or yellowish-white powder or crystalline powder, hygroscopic.

Solubility: freely soluble in water and in methanol, soluble in anhydrous ethanol.

IDENTIFICATION

- A. Specific optical rotation (see Tests).
- B. Infrared absorption spectrophotometry (2.2.24).

 Comparison: Ph. Eur. reference spectrum of pravastatin
- C. 1 mL of solution S (see Tests) gives reaction (a) of sodium (2.3.1).

TESTS

Solution S. Dissolve 1.00 g in *carbon dioxide-free water R* and dilute to 20.0 mL with the same solvent.

Appearance of solution. The solution is clear (2.2.1) and not more intensely coloured than reference solution BY₆ $(2.2.2, Method\ II)$.

Dilute 2.0 mL of solution S to 10.0 mL with water R.

pH (2.2.3): 7.2 to 9.0 for solution S.

Specific optical rotation (2.2.7): + 153 to + 159 (anhydrous substance).

Dilute 2.0 mL of solution S to 20.0 mL with water R.

Related substances. Liquid chromatography (2.2.29).

Solvent mixture: methanol R, water R (9:11 V/V).

Test solution (a). Dissolve 0.1000 g of the substance to be examined in the solvent mixture and dilute to 100.0 mL with the solvent mixture.

Test solution (b). Dilute 10.0 mL of test solution (a) to 100.0 mL with the solvent mixture.

Reference solution (a). Dissolve the contents of a vial of pravastatin impurity A CRS in 1.0 mL of test solution (b). Reference solution (b). Dilute 2.0 mL of test solution (a) to 100.0 mL with the solvent mixture. Dilute 1.0 mL of this solution to 10.0 mL with the solvent mixture.