#### **IDENTIFICATION**

- A. Relative density (see Tests).
- B. Refractive index (see Tests).
- C. Boiling point (2.2.12): 184 °C to 189 °C.
- D. To 0.5 mL add 5 mL of *pyridine R* and 2 g of finely ground *nitrobenzoyl chloride R*. Boil for 1 min and pour into 15 mL of cold *water R* with shaking. Filter, wash the precipitate with 20 mL of a saturated solution of *sodium hydrogen carbonate R* and then with *water R* and dry. Dissolve in boiling *ethanol (80 per cent V/V) R* and filter the hot solution. On cooling, crystals are formed which, after drying at 100-105 °C, melt (2.2.14) at 121 °C to 128 °C.

#### TESTS

Appearance. It is clear (2.2.1) and colourless (2.2.2, Method II).

**Relative density** (2.2.5): 1.035 to 1.040. **Refractive index** (2.2.6): 1.431 to 1.433.

**Acidity.** To 10 mL add 40 mL of *water R* and 0.1 mL of *bromothymol blue solution R1*. The solution is greenish-yellow. Not more than 0.05 mL of 0.1 M sodium hydroxide is required to change the colour of the indicator to blue.

**Oxidising substances.** To 10 mL add 5 mL of water R, 2 mL of potassium iodide solution R and 2 mL of dilute sulfuric acid R and allow to stand in a ground-glass-stoppered flask protected from light for 15 min. Titrate with  $0.05\,M$  sodium thiosulfate, using 1 mL of starch solution R as indicator. Not more than  $0.2\,\text{mL}$  of  $0.05\,M$  sodium thiosulfate is required.

**Reducing substances.** To 1 mL add 1 mL of *dilute ammonia R1* and heat in a water-bath at 60 °C for 5 min. The solution is not yellow. Immediately add 0.15 mL of 0.1 M silver nitrate and allow to stand for 5 min. The solution does not change its appearance.

**Heavy metals** (2.4.8). Mix 4 mL with 16 mL of water R. 12 mL of the solution complies with test A for heavy metals (5 ppm m/V). Prepare the reference solution using lead standard solution (1 ppm Pb) R.

**Water** (2.5.12). Not more than 0.2 per cent, determined on 5.00 g by the semi-micro determination of water.

**Sulfated ash** (2.4.14). Heat 50 g until it burns and ignite. Allow to cool. Moisten the residue with *sulfuric acid R* and ignite; repeat the operations. The residue weighs not more than 5 mg (0.01 per cent).

### STORAGE

Store in an airtight container.

01/2008:2122

# PROPYLENE GLYCOL DICAPRYLOCAPRATE

# Propylenglycoli dicaprylocapras

#### DEFINITION

Propylene glycol diesters of saturated fatty acids, mainly caprylic (octanoic) acid and capric (decanoic) acid, of vegetable origin.

#### **CHARACTERS**

*Appearance*: almost colourless to light yellow, oily liquid. *Solubility*: practically insoluble in water, soluble in fatty oils and in light petroleum, slightly soluble in anhydrous ethanol.

#### IDENTIFICATION

- A. Refractive index (2.2.6): 1.439 to 1.442.
- B. Relative density (2.2.5): 0.910 to 0.930.
- C. Viscosity (2.2.9): 9 mPa·s to 12 mPa·s.
- D. Composition of fatty acids (see Tests).

#### TESTS

**Appearance**. The substance to be examined is clear (2.2.1) and not more intensely coloured than reference solution BY<sub>6</sub>  $(2.2.2, Method\ II)$ .

Acid value (2.5.1): maximum 0.2.

**Hydroxyl value** (2.5.3, Method A): maximum 10.

**Iodine value** (2.5.4): maximum 1.0.

**Peroxide value** (2.5.5, Method A): maximum 1.0.

**Saponification value** (2.5.6): 320 to 340.

**Unsaponifiable matter** (2.5.7): maximum 0.3 per cent, determined on 5.0 g.

**Alkaline impurities.** Dissolve 2.00 g of the substance to be examined in a mixture of 1.5 mL of *ethanol* (*96 per cent*) *R* and 3.0 mL of *ether R*. Add 0.05 mL of *bromophenol blue solution R*. Not more than 0.15 mL of 0.01 *M hydrochloric acid* is required to change the colour of the indicator to yellow.

**Composition of fatty acids**. Gas chromatography (*2.4.22*, *Method C*). Prepare reference solution (a) as indicated in Table 2.4.22.-2.

#### Column:

- material: fused silica,
- size: l = 30 m,  $\emptyset = 0.32 \text{ mm}$ ,
- stationary phase: macrogol 20 000 R (film thickness

Carrier gas: helium for chromatography R.

Flow rate: 1.3 mL/min. Split ratio: 1:100. Temperature:

	Time (min)	Temperature (°C)
Column	0 - 1	70
	1 - 35	$70 \rightarrow 240$
	35 - 50	240
njection port		250
Detector		250

Detection: flame ionisation.

Composition of the fatty acid fraction of the substance to be examined:

- caproic acid: maximum 2.0 per cent,
- caprylic acid: 50.0 per cent to 80.0 per cent,
- capric acid: 20.0 per cent to 50.0 per cent,
- lauric acid: maximum 3.0 per cent,
- myristic acid: maximum 1.0 per cent.

Water (2.5.12): maximum 0.1 per cent, determined on 5.00 g.

Total ash (2.4.16): maximum 0.1 per cent, determined on 2.0 g.

STORAGE

Protected from light.

01/2008:2087

## PROPYLENE GLYCOL DILAURATE

## Propylenglycoli dilauras

## DEFINITION

Mixture of propylene glycol mono- and diesters of lauric (dodecanoic) acid.

*Content*: minimum 70.0 per cent of diesters and maximum 30.0 per cent of monoesters.