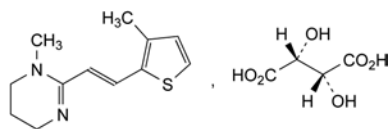


- I. 9,21-dichloro-11β-hydroxy-16α-methyl-3,20-dioxo-5ξ-pregn-1-ene-6ξ,17-diyl 6-acetate 17-(furan-2-carboxylate).

01/2008:1546  
corrected 6.0

## MORANTEL HYDROGEN TARTRATE FOR VETERINARY USE

### Moranteli hydrogenotartras ad usum veterinarium



$C_{16}H_{22}N_2O_6S$   
[26155-31-7]

$M_r$  370.4

#### DEFINITION

1-Methyl-2-[(*E*)-2-(3-methylthiophen-2-yl)ethenyl]-1,4,5,6-tetrahydropyrimidine hydrogen tartrate.

*Content*: 98.5 per cent to 101.5 per cent (dried substance).

#### CHARACTERS

*Appearance*: white or pale yellow, crystalline powder.

*Solubility*: very soluble in water and in ethanol (96 per cent), practically insoluble in ethyl acetate.

#### IDENTIFICATION

*First identification*: B.

*Second identification*: A, C, D.

A. Melting point (2.2.14): 167 °C to 172 °C.

B. Infrared absorption spectrophotometry (2.2.24).

*Comparison*: morantel hydrogen tartrate CRS.

C. Dissolve about 10 mg in 1 mL of a 5 g/L solution of ammonium vanadate R. Evaporate to dryness. Add 0.1 mL of sulfuric acid R. A purple colour is produced.

D. Dissolve about 10 mg in 1 mL of 0.1 M sodium hydroxide. Transfer to a separating funnel and shake with 5 mL of methylene chloride R. Discard the organic layer. Neutralise the aqueous layer with a few drops of dilute hydrochloric acid R. The solution gives reaction (b) of tartrates (2.3.1).

#### TESTS

**Solution S.** Dissolve 0.25 g in carbon dioxide-free water R and dilute to 25.0 mL with the same solvent.

**Appearance of solution.** Solution S is clear (2.2.1) and not more intensely coloured than reference solution GY<sub>6</sub> or Y<sub>6</sub> (2.2.2, Method II).

**pH** (2.2.3): 3.3 to 3.9 for solution S.

**Related substances.** Liquid chromatography (2.2.29). Carry out the test protected from light.

**Test solution.** Dissolve 50.0 mg of the substance to be examined in the mobile phase and dilute to 100.0 mL with the mobile phase.

**Reference solution (a).** Dilute 1.0 mL of the test solution to 100.0 mL with the mobile phase.

**Reference solution (b).** Dilute 2.0 mL of reference solution (a) to 100.0 mL with the mobile phase.

**Reference solution (c).** Expose 10 mL of reference solution (a) to daylight for 15 min before injection.

**Reference solution (d).** Dissolve 15.0 mg of tartaric acid R in the mobile phase and dilute to 100.0 mL with the mobile phase.

#### Column:

– size:  $l = 0.25$  m,  $\varnothing = 4.6$  mm;

– stationary phase: base-deactivated end-capped octadecylsilyl silica gel for chromatography R (5 µm).

**Mobile phase:** to a mixture of 0.35 volumes of triethylamine R and 85 volumes of water R adjusted to pH 2.5 with phosphoric acid R, add 5 volumes of tetrahydrofuran R and 10 volumes of methanol R.

**Flow rate:** 0.75 mL/min.

**Detection:** spectrophotometer at 226 nm.

**Injection:** 20 µL.

**Run time:** twice the retention time of morantel.

**System suitability:** reference solution (c):

– resolution: minimum of 2 between the principal peak and the preceding peak ((*Z*)-isomer).

#### Limits:

– any impurity apart from the peak due to tartaric acid: not more than 0.5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.5 per cent);

– total: not more than the area of the principal peak in the chromatogram obtained with reference solution (a) (1 per cent);

– disregard limit: the area of the principal peak in the chromatogram obtained with reference solution (b) (0.02 per cent).

**Heavy metals** (2.4.8): maximum 20 ppm.

1.0 g complies with test C. Prepare the reference solution using 2 mL of lead standard solution (10 ppm Pb) R.

**Loss on drying** (2.2.32): maximum 1.5 per cent, determined on 1.000 g by drying in an oven at 105 °C.

**Sulfated ash** (2.4.14): maximum 0.1 per cent, determined on 1.0 g.

#### ASSAY

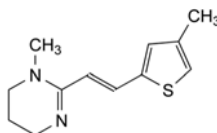
Dissolve 0.280 g in 40 mL of anhydrous acetic acid R. Titrate with 0.1 M perchloric acid, determining the end-point potentiometrically (2.2.20).

1 mL of 0.1 M perchloric acid is equivalent to 37.04 mg of  $C_{16}H_{22}N_2O_6S$ .

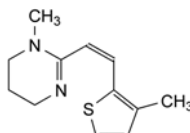
#### STORAGE

Protected from light.

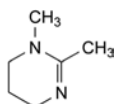
#### IMPURITIES



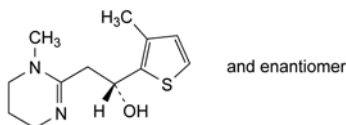
- A. 1-methyl-2-[(*E*)-2-(4-methylthiophen-2-yl)ethenyl]-1,4,5,6-tetrahydropyrimidine,



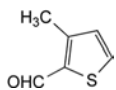
- B. 1-methyl-2-[(*Z*)-2-(3-methylthiophen-2-yl)ethenyl]-1,4,5,6-tetrahydropyrimidine,



C. 1,2-dimethyl-1,4,5,6-tetrahydropyrimidine,



D. (1*RS*)-2-(1-methyl-1,4,5,6-tetrahydropyrimidin-2-yl)-1-(3-methylthiophen-2-yl)ethanol,

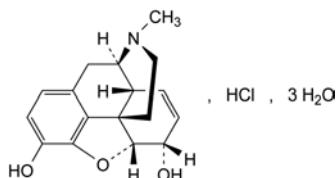


E. 3-methylthiophene-2-carbaldehyde.

04/2008:0097  
corrected 6.7

## MORPHINE HYDROCHLORIDE

### Morphini hydrochloridum



$C_{17}H_{20}ClNO_3 \cdot 3H_2O$   
[6055-06-7]

$M_r$  375.8

#### DEFINITION

7,8-Didehydro-4,5 $\alpha$ -epoxy-17-methylmorphinan-3,6 $\alpha$ -diol hydrochloride trihydrate.

*Content*: 98.0 per cent to 102.0 per cent (anhydrous substance).

#### CHARACTERS

*Appearance*: white or almost white, crystalline powder or colourless, silky needles or cubical masses, efflorescent in a dry atmosphere.

*Solubility*: soluble in water, slightly soluble in ethanol (96 per cent), practically insoluble in toluene.

#### IDENTIFICATION

*First identification*: A, E.

*Second identification*: B, C, D, E.

A. Infrared absorption spectrophotometry (2.2.24).

*Comparison*: morphine hydrochloride CRS.

B. Ultraviolet and visible absorption spectrophotometry (2.2.25).

*Solution A*. Dissolve 25.0 mg in water R and dilute to 25.0 mL with the same solvent.

*Test solution (a)*. Dilute 10.0 mL of solution A to 100.0 mL with water R.

*Test solution (b)*. Dilute 10.0 mL of solution A to 100.0 mL with 0.1 M sodium hydroxide.

*Spectral range*: 250-350 nm for test solutions (a) and (b).

*Absorption maximum*: at 285 nm for test solution (a); at 298 nm for test solution (b).

*Specific absorbance at the absorption maximum*: 37 to 43 for test solution (a); 64 to 72 for test solution (b).

C. To about 1 mg of powdered substance in a porcelain dish add 0.5 mL of sulfuric acid-formaldehyde reagent R. A purple colour develops and becomes violet.

D. It gives the reaction of alkaloids (2.3.1).

E. It gives reaction (a) of chlorides (2.3.1).

#### TESTS

**Solution S**. Dissolve 0.500 g in carbon dioxide-free water R and dilute to 25.0 mL with the same solvent.

**Appearance of solution**. Solution S is clear (2.2.1) and not more intensely coloured than reference solution Y<sub>6</sub> or BY<sub>6</sub> (2.2.2, Method II).

**Acidity or alkalinity**. To 10 mL of solution S add 0.05 mL of methyl red solution R. Not more than 0.2 mL of 0.02 M sodium hydroxide or 0.02 M hydrochloric acid is required to change the colour of the indicator.

**Specific optical rotation (2.2.7)**: –110 to –115 (anhydrous substance), determined on solution S.

**Related substances**. Liquid chromatography (2.2.29).

*Test solution*. Dissolve 0.125 g of the substance to be examined in a 1 per cent V/V solution of acetic acid R and dilute to 50 mL with the same solution.

*Reference solution (a)*. Dilute 1.0 mL of the test solution to 100.0 mL with a 1 per cent V/V solution of acetic acid R. Dilute 2.0 mL of this solution to 10.0 mL with a 1 per cent V/V solution of acetic acid R.

*Reference solution (b)*. Dissolve 5 mg of morphine for system suitability CRS (containing impurities B, C, E and F) in a 1 per cent V/V solution of acetic acid R and dilute to 2 mL with the same solution.

*Column*:

– size:  $l = 0.15$  m,  $\varnothing = 4.6$  mm;

– stationary phase: end-capped octadecylsilyl silica gel for chromatography R (5  $\mu$ m);

– temperature: 35 °C.

*Mobile phase*:

– mobile phase A: 1.01 g/L solution of sodium heptanesulfonate R adjusted to pH 2.6 with a 50 per cent V/V solution of phosphoric acid R;

– mobile phase B: methanol R;

Time (min)	Mobile phase A (per cent V/V)	Mobile phase B (per cent V/V)
0 - 2	85	15
2 - 35	85 → 50	15 → 50
35 - 40	50	50

*Flow rate*: 1.5 mL/min.

*Detection*: spectrophotometer at 230 nm.

*Injection*: 10  $\mu$ L.

*Identification of impurities*: use the chromatogram supplied with morphine for system suitability CRS and the chromatogram obtained with reference solution (b) to identify the peaks due to impurities B, C, E and F.

*Relative retention* with reference to morphine (retention time = about 12.5 min): impurity F = about 0.95; impurity E = about 1.1; impurity C = about 1.6; impurity B = about 1.9.

*System suitability*: reference solution (b):

– peak-to-valley ratio: minimum 2, where  $H_p$  = height above the baseline of the peak due to impurity F and  $H_v$  = height above the baseline of the lowest point of the curve separating this peak from the peak due to morphine.

*Limits*:

– correction factors: for the calculation of content, multiply the peak areas of the following impurities by the corresponding correction factor: impurity B = 0.25; impurity C = 0.4; impurity E = 0.5;

– impurity B: not more than twice the area of the principal peak in the chromatogram obtained with reference solution (a) (0.4 per cent);