# 07/2010:10000 corrected 7.0

# 1. GENERAL NOTICES

#### 1.1. GENERAL STATEMENTS

The General Notices apply to all monographs and other texts of the European Pharmacopoeia.

The official texts of the European Pharmacopoeia are published in English and French. Translations in other languages may be prepared by the signatory States of the European Pharmacopoeia Convention. In case of doubt or dispute, the English and French versions are alone authoritative.

In the texts of the European Pharmacopoeia, the word 'Pharmacopoeia' without qualification means the European Pharmacopoeia. The official abbreviation Ph. Eur. may be used to indicate the European Pharmacopoeia.

The use of the title or the subtitle of a monograph implies that the article complies with the requirements of the relevant monograph. Such references to monographs in the texts of the Pharmacopoeia are shown using the monograph title and reference number in *italics*.

A preparation must comply throughout its period of validity; a distinct period of validity and/or specifications for opened or broached containers may be decided by the competent authority. The subject of any other monograph must comply throughout its period of use. The period of validity that is assigned to any given article and the time from which that period is to be calculated are decided by the competent authority in light of experimental results of stability studies.

Unless otherwise indicated in the General Notices or in the monographs, statements in monographs constitute mandatory requirements. General chapters become mandatory when referred to in a monograph, unless such reference is made in a way that indicates that it is not the intention to make the text referred to mandatory but rather to cite it for information.

The active substances, excipients, pharmaceutical preparations and other articles described in the monographs are intended for human and veterinary use (unless explicitly restricted to one of these uses). An article is not of Pharmacopoeia quality unless it complies with all the requirements stated in the monograph. This does not imply that performance of all the tests in a monograph is necessarily a prerequisite for a manufacturer in assessing compliance with the Pharmacopoeia before release of a product. The manufacturer may obtain assurance that a product is of Pharmacopoeia quality from data derived, for example, from validation studies of the manufacturing process and from in-process controls. Parametric release in circumstances deemed appropriate by the competent authority is thus not precluded by the need to comply with the Pharmacopoeia.

The tests and assays described are the official methods upon which the standards of the Pharmacopoeia are based. With the agreement of the competent authority, alternative methods of analysis may be used for control purposes, provided that the methods used enable an unequivocal decision to be made as to whether compliance with the standards of the monographs would be achieved if the official methods were used. In the event of doubt or dispute, the methods of analysis of the Pharmacopoeia are alone authoritative.

Certain materials that are the subject of a pharmacopoeial monograph may exist in different grades suitable for different purposes. Unless otherwise indicated in the monograph, the requirements apply to all grades of the material. In some monographs, particularly those on excipients, a list of functionality-related characteristics that are relevant to the use of the substance may be appended to the monograph for information. Test methods for determination of one or more of these characteristics may be given, also for information.

**Quality systems**. The quality standards represented by monographs are valid only where the articles in question are produced within the framework of a suitable quality system.

**General monographs.** Substances and preparations that are the subject of an individual monograph are also required to comply with relevant, applicable general monographs. Cross-references to applicable general monographs are not normally given in individual monographs.

General monographs apply to all substances and preparations within the scope of the Definition section of the general monograph, except where a preamble limits the application, for example to substances and preparations that are the subject of a monograph of the Pharmacopoeia.

General monographs on dosage forms apply to all preparations of the type defined. The requirements are not necessarily comprehensive for a given specific preparation and requirements additional to those prescribed in the general monograph may be imposed by the competent authority.

General monographs and individual monographs are complementary. If the provisions of a general monograph do not apply to a particular product, this is expressly stated in the individual monograph.

Validation of pharmacopoeial methods. The test methods given in monographs and general chapters have been validated in accordance with accepted scientific practice and current recommendations on analytical validation. Unless otherwise stated in the monograph or general chapter, validation of the test methods by the analyst is not required.

**Conventional terms**. The term 'competent authority' means the national, supranational or international body or organisation vested with the authority for making decisions concerning the issue in question. It may, for example, be a national pharmacopoeia authority, a licensing authority or an official control laboratory.

The expression 'unless otherwise justified and authorised' means that the requirements have to be met, unless the competent authority authorises a modification or an exemption where justified in a particular case.

Statements containing the word 'should' are informative or advisory.

In certain monographs or other texts, the terms 'suitable' and 'appropriate' are used to describe a reagent, micro-organism, test method etc.; if criteria for suitability are not described in the monograph, suitability is demonstrated to the satisfaction of the competent authority.

Medicinal product. (a) Any substance or combination of substances presented as having properties for treating or preventing disease in human beings and/or animals; or (b) any substance or combination of substances that may be used in or administered to human beings and/or animals with a view either to restoring, correcting or modifying physiological functions by exerting a pharmacological, immunological or metabolic action, or to making a medical diagnosis.

Herbal medicinal product. Any medicinal product, exclusively containing as active ingredients one or more herbal drugs or one or more herbal drug preparations, or one or more such herbal drugs in combination with one or more such herbal drug preparations.

Active substance. Any substance intended to be used in the manufacture of a medicinal product and that, when so used, becomes an active ingredient of the medicinal product. Such substances are intended to furnish a pharmacological activity or other direct effect in the diagnosis, cure, mitigation, treatment or prevention of disease, or to affect the structure and function of the body.

*Excipient* (auxiliary substance). Any constituent of a medicinal product that is not an active substance. Adjuvants, stabilisers, antimicrobial preservatives, diluents, antioxidants, for example, are excipients.

Interchangeable methods. Certain general chapters contain a statement that the text in question is harmonised with the corresponding text of the Japanese Pharmacopoeia and/or the United States Pharmacopeia and that these texts are interchangeable. This implies that if a substance or preparation is found to comply with a requirement using an interchangeable method from one of these pharmacopoeias it complies with the requirements of the European Pharmacopoeia. In the event of doubt or dispute, the text of the European Pharmacopoeia is alone authoritative.

References to regulatory documents. Monographs and general chapters may contain references to documents issued by regulatory authorities for medicines, for example directives and notes for guidance of the European Union. These references are provided for information for users for the Pharmacopoeia. Inclusion of such a reference does not modify the status of the documents referred to, which may be mandatory or for guidance.

# 1.2. OTHER PROVISIONS APPLYING TO GENERAL CHAPTERS AND MONOGRAPHS

**Quantities.** In tests with numerical limits and assays, the quantity stated to be taken for examination is approximate. The amount actually used, which may deviate by not more than 10 per cent from that stated, is accurately weighed or measured and the result is calculated from this exact quantity. In tests where the limit is not numerical, but usually depends upon comparison with the behaviour of a reference substance in the same conditions, the stated quantity is taken for examination. Reagents are used in the prescribed amounts.

Quantities are weighed or measured with an accuracy commensurate with the indicated degree of precision. For weighings, the precision corresponds to plus or minus 5 units after the last figure stated (for example, 0.25 g is to be interpreted as 0.245 g to 0.255 g). For the measurement of volumes, if the figure after the decimal point is a zero or ends in a zero (for example, 10.0 mL or 0.50 mL), the volume is measured using a pipette, a volumetric flask or a burette, as appropriate; otherwise, a graduated measuring cylinder or a graduated pipette may be used. Volumes stated in microlitres are measured using a micropipette or microsyringe.

It is recognised, however, that in certain cases the precision with which quantities are stated does not correspond to the number of significant figures stated in a specified numerical limit. The weighings and measurements are then carried out with a sufficiently improved accuracy.

**Apparatus and procedures.** Volumetric glassware complies with Class A requirements of the appropriate International Standard issued by the International Organisation for Standardisation.

Unless otherwise prescribed, analytical procedures are carried out at a temperature between 15  $^{\circ}\text{C}$  and 25  $^{\circ}\text{C}.$ 

Unless otherwise prescribed, comparative tests are carried out using identical tubes of colourless, transparent, neutral glass with a flat base; the volumes of liquid prescribed are for use with tubes having an internal diameter of 16 mm, but tubes with a larger internal diameter may be used provided the volume of liquid used is adjusted (2.1.5). Equal volumes of the liquids to be compared are examined down the vertical axis of the tubes against a white background, or if necessary against a black background. The examination is carried out in diffuse light.

Any solvent required in a test or assay in which an indicator is to be used is previously neutralised to the indicator, unless a blank test is prescribed.

**Water-bath**. The term 'water-bath' means a bath of boiling water unless water at another temperature is indicated. Other methods of heating may be substituted provided the temperature is near to but not higher than 100 °C or the indicated temperature.

**Drying and ignition to constant mass**. The terms 'dried to constant mass' and 'ignited to constant mass' mean that 2 consecutive weighings do not differ by more than 0.5 mg, the  $2^{nd}$  weighing following an additional period of drying or of ignition respectively appropriate to the nature and quantity of the residue.

Where drying is prescribed using one of the expressions 'in a desiccator' or 'in vacuo', it is carried out using the conditions described in chapter 2.2.32. Loss on drying.

**Reagents**. The proper conduct of the analytical procedures described in the Pharmacopoeia and the reliability of the results depend, in part, upon the quality of the reagents used. The reagents are described in general chapter 4. It is assumed that reagents of analytical grade are used; for some reagents, tests to determine suitability are included in the specifications.

**Solvents.** Where the name of the solvent is not stated, the term 'solution' implies a solution in water.

Where the use of water is specified or implied in the analytical procedures described in the Pharmacopoeia or for the preparation of reagents, water complying with the requirements of the monograph *Purified water (0008)* is used, except that for many purposes the requirements for bacterial endotoxins (*Purified water in bulk*) and microbial contamination (*Purified water in containers*) are not relevant. The term 'distilled water' indicates purified water prepared by distillation.

The term 'ethanol' without qualification means anhydrous ethanol. The term 'alcohol' without qualification means ethanol (96 per cent). Other dilutions of ethanol are indicated by the term 'ethanol' or 'alcohol' followed by a statement of the percentage by volume of ethanol ( $C_0H_6O$ ) required.

**Expression of content.** In defining content, the expression 'per cent' is used according to circumstances with one of 2 meanings:

- per cent m/m (percentage, mass in mass) expresses the number of grams of substance in 100 grams of final product;
- per cent V/V (percentage, volume in volume) expresses the number of millilitres of substance in 100 mL of final product.

The expression 'parts per million' (or ppm) refers to mass in mass, unless otherwise specified.

**Temperature.** Where an analytical procedure describes temperature without a figure, the general terms used have the following meaning:

in a deep-freeze: below – 15 °C;
in a refrigerator: 2 °C to 8 °C;
cold or cool: 8 °C to 15 °C;
room temperature: 15 °C to 25 °C.

## 1.3. GENERAL CHAPTERS

Containers. Materials used for containers are described in general chapter 3.1. General names used for materials, particularly plastic materials, each cover a range of products varying not only in the properties of the principal constituent but also in the additives used. The test methods and limits for materials depend on the formulation and are therefore applicable only for materials whose formulation is covered by the preamble to the specification. The use of materials with different formulations, and the test methods and limits applied to them, are subject to agreement by the competent authority.

The specifications for containers in general chapter 3.2 have been developed for general application to containers of the stated category, but in view of the wide variety of containers available and possible new developments, the publication of a specification does not exclude the use, in justified circumstances, of containers that comply with other specifications, subject to agreement by the competent authority. Reference may be made within the monographs of the Pharmacopoeia to the definitions and specifications for containers provided in chapter 3.2. Containers. The general monographs for pharmaceutical dosage forms may, under the

heading Definition/Production, require the use of certain types of container; certain other monographs may, under the heading Storage, indicate the type of container that is recommended for use.

#### 1.4. MONOGRAPHS

#### TITLES

Monograph titles are in English and French in the respective versions and there is a Latin subtitle.

## RELATIVE ATOMIC AND MOLECULAR MASSES

The relative atomic mass  $(A_r)$  or the relative molecular mass  $(M_r)$  is shown, as and where appropriate, at the beginning of each monograph. The relative atomic and molecular masses and the molecular and graphic formulae do not constitute analytical standards for the substances described.

# CHEMICAL ABSTRACTS SERVICE (CAS) REGISTRY NUMBER

CAS registry numbers are included for information in monographs, where applicable, to provide convenient access to useful information for users. CAS Registry Number® is a Registered Trademark of the American Chemical Society.

#### DEFINITION

Statements under the heading Definition constitute an official definition of the substance, preparation or other article that is the subject of the monograph.

**Limits of content.** Where limits of content are prescribed, they are those determined by the method described under Assay.

**Herbal drugs**. In monographs on herbal drugs, the definition indicates whether the subject of the monograph is, for example, the whole drug or the drug in powdered form. Where a monograph applies to the drug in several states, for example both to the whole drug and the drug in powdered form, the definition states this.

#### PRODUCTION

Statements under the heading Production draw attention to particular aspects of the manufacturing process but are not necessarily comprehensive. They constitute mandatory requirements for manufacturers, unless otherwise stated. They may relate, for example, to source materials; to the manufacturing process itself and its validation and control; to in-process testing; or to testing that is to be carried out by the manufacturer on the final article, either on selected batches or on each batch prior to release. These statements cannot necessarily be verified on a sample of the final article by an independent analyst. The competent authority may establish that the instructions have been followed, for example, by examination of data received from the manufacturer, by inspection of manufacture or by testing appropriate samples.

The absence of a Production section does not imply that attention to features such as those referred to above is not required.

# Choice of vaccine strain, Choice of vaccine composition.

The Production section of a monograph may define the characteristics of a vaccine strain or vaccine composition. Unless otherwise stated, test methods given for verification of these characteristics are provided for information as examples of suitable methods. Subject to approval by the competent authority, other test methods may be used without validation against the method shown in the monograph.

#### CHARACTERS

The statements under the heading Characters are not to be interpreted in a strict sense and are not requirements.

**Solubility**. In statements of solubility in the Characters section, the terms used have the following significance, referred to a temperature between 15  $^{\circ}$ C and 25  $^{\circ}$ C.

Descriptive term	Approximate volume of solvent in millilitres per gram of solute				
Very soluble	less than	1			
Freely soluble	from	1	to	10	
Soluble	from	10	to	30	
Sparingly soluble	from	30	to	100	
Slightly soluble	from	100	to	1000	
Very slightly soluble	from	1000	to	10 000	
Practically insoluble	more than			10 000	

The term 'partly soluble' is used to describe a mixture where only some of the components dissolve. The term 'miscible' is used to describe a liquid that is miscible in all proportions with the stated solvent.

#### IDENTIFICATION

**Scope**. The tests given in the Identification section are not designed to give a full confirmation of the chemical structure or composition of the product; they are intended to give confirmation, with an acceptable degree of assurance, that the article conforms to the description on the label.

First and second identifications. Certain monographs have subdivisions entitled 'First identification' and 'Second identification'. The test or tests that constitute the 'First identification' may be used in all circumstances. The test or tests that constitute the 'Second identification' may be used in pharmacies provided it can be demonstrated that the substance or preparation is fully traceable to a batch certified to comply with all the other requirements of the monograph.

Certain monographs give two or more sets of tests for the purpose of the first identification, which are equivalent and may be used independently. One or more of these sets usually contain a cross-reference to a test prescribed in the Tests section of the monograph. It may be used to simplify the work of the analyst carrying out the identification and the prescribed tests. For example, one identification set cross-refers to a test for enantiomeric purity while the other set gives a test for specific optical rotation: the intended purpose of the two is the same, that is, verification that the correct enantiomer is present.

**Powdered herbal drugs.** Monographs on herbal drugs may contain schematic drawings of the powdered drug. These drawings complement the description given in the relevant identification test.

# TESTS AND ASSAYS

**Scope**. The requirements are not framed to take account of all possible impurities. It is not to be presumed, for example, that an impurity that is not detectable by means of the prescribed tests is tolerated if common sense and good pharmaceutical practice require that it be absent. See also below under Impurities.

**Calculation.** Where the result of a test or assay is required to be calculated with reference to the dried or anhydrous substance or on some other specified basis, the determination of loss on drying, water content or other property is carried out by the method prescribed in the relevant test in the monograph. The words 'dried substance' or 'anhydrous substance' etc. appear in parentheses after the result.

**Limits**. The limits prescribed are based on data obtained in normal analytical practice; they take account of normal analytical errors, of acceptable variations in manufacture and compounding and of deterioration to an extent considered

acceptable. No further tolerances are to be applied to the limits prescribed to determine whether the article being examined complies with the requirements of the monograph.

In determining compliance with a numerical limit, the calculated result of a test or assay is first rounded to the number of significant figures stated, unless otherwise prescribed. The limits, regardless of whether the values are expressed as percentages or as absolute values, are considered significant to the last digit shown (for example 140 indicates 3 significant figures). The last figure of the result is increased by one when the part rejected is equal to or exceeds one half-unit, whereas it is not modified when the part rejected is less than a half-unit.

Indication of permitted limit of impurities. For comparative tests, the approximate content of impurity tolerated, or the sum of impurities, may be indicated for information only. Acceptance or rejection is determined on the basis of compliance or non-compliance with the stated test. If the use of a reference substance for the named impurity is not prescribed, this content may be expressed as a nominal concentration of the substance used to prepare the reference solution specified in the monograph, unless otherwise described.

**Herbal drugs**. For herbal drugs, the sulfated ash, total ash, water-soluble matter, alcohol-soluble matter, water content, content of essential oil and content of active principle are calculated with reference to the drug that has not been specially dried, unless otherwise prescribed in the monograph.

**Equivalents.** Where an equivalent is given, for the purposes of the Pharmacopoeia only the figures shown are to be used in applying the requirements of the monograph.

**Culture media.** The culture media described in monographs and general chapters have been found to be satisfactory for the intended purpose. However, the components of media, particularly those of biological origin, are of variable quality, and it may be necessary for optimal performance to modulate the concentration of some ingredients, notably:

- peptones and meat or yeast extracts, with respect to their nutritive properties;
- buffering substances;
- bile salts, bile extract, deoxycholate, and colouring matter, depending on their selective properties;
- antibiotics, with respect to their activity.

## **STORAGE**

The information and recommendations given under the heading Storage do not constitute a pharmacopoeial requirement but the competent authority may specify particular storage conditions that must be met.

The articles described in the Pharmacopoeia are stored in such a way as to prevent contamination and, as far as possible, deterioration. Where special conditions of storage are recommended, including the type of container (see section 1.3. General chapters) and limits of temperature, they are stated in the monograph.

The following expressions are used in monographs under Storage with the meaning shown.

In an airtight container means that the product is stored in an airtight container (3.2). Care is to be taken when the container is opened in a damp atmosphere. A low moisture content may be maintained, if necessary, by the use of a desiccant in the container provided that direct contact with the product is avoided.

Protected from light means that the product is stored either in a container made of a material that absorbs actinic light sufficiently to protect the contents from change induced by such light, or in a container enclosed in an outer cover that provides such protection, or is stored in a place from which all such light is excluded.

#### LABELLING

In general, labelling of medicines is subject to supranational and national regulation and to international agreements. The statements under the heading Labelling are not therefore comprehensive and, moreover, for the purposes of the Pharmacopoeia only those statements that are necessary to demonstrate compliance or non-compliance with the monograph are mandatory. Any other labelling statements are included as recommendations. When the term 'label' is used in the Pharmacopoeia, the labelling statements may appear on the container, the package, a leaflet accompanying the package, or a certificate of analysis accompanying the article, as decided by the competent authority.

#### WARNINGS

Materials described in monographs and reagents specified for use in the Pharmacopoeia may be injurious to health unless adequate precautions are taken. The principles of good quality control laboratory practice and the provisions of any appropriate regulations are to be observed at all times. Attention is drawn to particular hazards in certain monographs by means of a warning statement; absence of such a statement is not to be taken to mean that no hazard exists.

#### **IMPURITIES**

A list of all known and potential impurities that have been shown to be detected by the tests in a monograph may be given. See also chapter 5.10. Control of impurities in substances for pharmaceutical use. The impurities are designated by a letter or letters of the alphabet. Where a letter appears to be missing, the impurity designated by this letter has been deleted from the list during monograph development prior to publication or during monograph revision.

# FUNCTIONALITY-RELATED CHARACTERISTICS OF EXCIPIENTS

Monographs on excipients may have a section on functionality-related characteristics. The characteristics, any test methods for determination and any tolerances are not mandatory requirements; they may nevertheless be relevant for use of the excipient and are given for information (see also section 1.1. General statements).

# REFERENCE STANDARDS

Certain monographs require the use of reference standards (chemical reference substances, biological reference preparations, reference spectra). See also chapter 5.12. Reference standards. The European Pharmacopoeia Commission establishes the official reference standards, which are alone authoritative in case of arbitration. These reference standards are available from the European Directorate for the Quality of Medicines & HealthCare (EDQM). Information on the available reference standards and a batch validity statement can be obtained via the EDQM website.

## 1.5. ABBREVIATIONS AND SYMBOLS

A	Absorbance
$A_{1~{ m cm}}^{1~{ m per~cent}}$	Specific absorbance
$A_{\mathrm{r}}$	Relative atomic mass
$[lpha]_{ m D}^{20}$	Specific optical rotation
bp	Boiling point
BRP	Biological reference preparation
CRS	Chemical reference substance
$d_{20}^{20}$	Relative density
λ	Wavelength
HRS	Herbal reference standard
IU	International Unit
M	Molarity

$M_{\rm r}$	Relative molecular mass	$PD_{50}$	The statistically determined dose of a vaccine that, in the conditions of the test, may
mp	Melting point		be expected to protect 50 per cent of the
$n_{ m D}^{20}$	Refractive index		animals against a challenge dose of the
Ph. Eur. U.	European Pharmacopoeia Unit		micro-organisms or toxins against which it is active
ppm	Parts per million (milligrams per kilogram)	$\mathrm{ED}_{50}$	The statistically determined dose of a vaccine
R	Substance or solution defined under 4. Reagents	30	that, in the conditions of the test, may be expected to induce specific antibodies in
$R_F$	Retardation factor (see chapter 2.2.46)		50 per cent of the animals for the relevant vaccine antigens
$R_{st}$	Used in chromatography to indicate the ratio	PFU	Pock-forming units or plaque-forming units
	of the distance travelled by a substance to the distance travelled by a reference substance	SPF	Specified-pathogen-free
RV	Substance used as a primary standard in volumetric analysis (chapter 4.2.1)		micro-organisms
Abbreviations	used in the monographs on immunoglobulins,	ATCC	American Type Culture Collection 10801 University Boulevard
immunosera a			Manassas, Virginia 20110-2209, USA
$\mathrm{LD}_{50}$	The statistically determined quantity of a	C.I.P.	Collection de Bactéries de l'Institut Pasteur
	substance that, when administered by the specified route, may be expected to cause	C.I.I .	B.P. 52, 25 rue du Docteur Roux
	the death of 50 per cent of the test animals		75724 Paris Cedex 15, France
	within a given period	IMI	International Mycological Institute
MLD	Minimum lethal dose		Bakeham Lane
L+/10 dose	The smallest quantity of a toxin that, in the		Surrey TW20 9TY, Great Britain
	conditions of the test, when mixed with 0.1 IU of antitoxin and administered by the	I.P.	Collection Nationale de Culture de Microorganismes (C.N.C.M.)
	specified route, causes the death of the test animals within a given period		Institut Pasteur
L+ dose	The smallest quantity of a toxin that, in the		25, rue du Docteur Roux
L' dosc	conditions of the test, when mixed with 1 IU		75724 Paris Cedex 15, France
	of antitoxin and administered by the specified route, causes the death of the test animals	NCIMB	National Collection of Industrial and Marine Bacteria Ltd
1 /100 1	within a given period		23 St Machar Drive
lr/100 dose	The smallest quantity of a toxin that, in the conditions of the test, when mixed		Aberdeen AB2 1RY, Great Britain
	with 0.01 IU of antitoxin and injected	NCPF	National Collection of Pathogenic Fungi
	intracutaneously causes a characteristic reaction at the site of injection within a given		London School of Hygiene and Tropical Medicine
I /10 -1	period		Keppel Street
Lp/10 dose	The smallest quantity of toxin that, in the conditions of the test, when mixed with		London WC1E 7HT, Great Britain
	0.1 IU of antitoxin and administered by the	NCTC	National Collection of Type Cultures
	specified route, causes paralysis in the test animals within a given period		Central Public Health Laboratory
Lo/10 dose	The largest quantity of a toxin that, in the		Colindale Avenue
Lo/ 10 dose	conditions of the test, when mixed with 0.1 IU	Nava	London NW9 5HT, Great Britain
	of antitoxin and administered by the specified	NCYC	National Collection of Yeast Cultures
	route, does not cause symptoms of toxicity in the test animals within a given period		AFRC Food Research Institute
Lf dose	The quantity of toxin or toxoid that		Colney Lane
2. 4000	flocculates in the shortest time with 1 IU of	NITE	Norwich NR4 7UA, Great Britain Biological Resource Center
	antitoxin	MIL	Department of Biotechnology
$CCID_{50}$	The statistically determined quantity of virus that may be expected to infect 50 per cent of		National Institute of Technology and
	the cell cultures to which it is added		Evaluation
$\mathrm{EID}_{50}$	The statistically determined quantity of virus that may be expected to infect 50 per cent of		2-5-8 Kazusakamatari, Kisarazu-shi, Chiba, 292-0818
	the fertilised eggs into which it is inoculated		Japan
${ m ID}_{50}$	The statistically determined quantity of a	S.S.I.	Statens Serum Institut
	virus that may be expected to infect 50 per		80 Amager Boulevard, Copenhagen,
	cent of the animals into which it is inoculated		Denmark

# 1.6. UNITS OF THE INTERNATIONAL SYSTEM (SI) USED IN THE PHARMACOPOEIA AND EQUIVALENCE WITH OTHER UNITS

#### INTERNATIONAL SYSTEM OF UNITS (SI)

The International System of Units comprises 3 classes of units, namely base units, derived units and supplementary units<sup>(1)</sup>. The base units and their definitions are set out in Table 1.6-1.

The derived units may be formed by combining the base units according to the algebraic relationships linking the corresponding quantities. Some of these derived units have special names and symbols. The SI units used in the Pharmacopoeia are shown in Table 1.6-2.

Some important and widely used units outside the International System are shown in Table 1.6-3.

The prefixes shown in Table 1.6-4 are used to form the names and symbols of the decimal multiples and submultiples of SI units.

#### NOTES

1. In the Pharmacopoeia, the Celsius temperature is used (symbol *t*). This is defined by the following equation:

$$t = T - T_0$$

where  $T_0$  = 273.15 K by definition. The Celsius or centigrade temperature is expressed in degrees Celsius (symbol °C). The unit 'degree Celsius' is equal to the unit 'kelvin'.

- 2. The practical expressions of concentrations used in the Pharmacopoeia are defined in the General Notices.
- 3. The radian is the plane angle between two radii of a circle that cut off on the circumference an arc equal in length to the radius.
- 4. In the Pharmacopoeia, conditions of centrifugation are defined by reference to the acceleration due to gravity (*q*):

$$g = 9.806 65 m \cdot s^{-2}$$

- 5. Certain quantities without dimensions are used in the Pharmacopoeia: relative density (2.2.5), absorbance (2.2.25), specific absorbance (2.2.25) and refractive index (2.2.6).
- 6. The microkatal is defined as the enzymic activity that, under defined conditions, produces the transformation (e.g. hydrolysis) of 1 micromole of the substrate per second.

Table 1.6.-1. - SI base units

Quantity		Unit		Definition	
Name	Symbol	Name	Symbol		
Length	1	metre	m	The metre is the length of the path travelled by light in a vacuum during a time interval of $1/299792458$ of a second.	
Mass	m	kilogram	kg	The kilogram is equal to the mass of the international prototype of the kilogram.	
Time	t	second	s	The second is the duration of 9 192 631 770 periods of the radiation corresponding to the transition between the two hyperfine levels of the ground state of the caesium-133 atom.	
Electric current	I	ampere	A	The ampere is that constant current which, maintained in two straight parallel conductors of infinite length, of negligible circular cross-section and placed 1 metre apart in vacuum would produce between these conductors a force equal to $2\times 10^{-7}$ newton per metre of length.	
Thermodynamic temperature	T	kelvin	K	The kelvin is the fraction $1/273.16$ of the thermodynamic temperature of the triple point of water.	
Amount of substance	n	mole	mol	The mole is the amount of substance of a system containing as many elementary entities as there are atoms in 0.012 kilogram of carbon-12*.	
Luminous intensity	$I_v$	candela	cd	The candela is the luminous intensity in a given direction of a source emitting monochromatic radiation with a frequency of $540 \times 10^{12}$ hertz and whose energy intensity in that direction is $1/683$ watt per steradian.	

<sup>\*</sup> When the mole is used, the elementary entities must be specified and may be atoms, molecules, ions, electrons, other particles or specified groups of such particles.

<sup>(1)</sup> The definitions of the units used in the International System are given in the booklet 'Le Système International d'Unités (SI)', published by the Bureau International des Poids et Mesures, Pavillon de Breteuil, F-92310 Sèvres.

Table 1.6.-2. – SI units used in the European Pharmacopoeia and equivalence with other units

Quantity				Unit		
Name	Symbol	Name	Symbol	Expression in SI base units	Expression in other SI units	Conversion of other units into SI units
Wave number	ν	one per metre	1/m	m <sup>-1</sup>		
Wavelength	λ	micrometre	µm nm	10 <sup>-6</sup> m		
Area	A, S	nanometre square metre	m <sup>2</sup>	10 <sup>-9</sup> m m <sup>2</sup>		
Volume	V	cubic metre	$\mathrm{m}^3$	m <sup>3</sup>		$1 \text{ mL} = 1 \text{ cm}^3 = 10^{-6} \text{ m}^3$
Frequency	v	hertz	Hz	s <sup>-1</sup>		
Density	ρ	kilogram per cubic metre	kg/m³	kg·m⁻³		$1 \text{ g/mL} = 1 \text{ g/cm}^3 = 10^3 \text{ kgm}^{-3}$
Velocity	v	metre per second	m/s	m·s⁻¹		
Force	F	newton	N	m·kg·s <sup>-2</sup>		1 dyne = 1 g·cm·s <sup>-2</sup> = 10 <sup>-5</sup> N 1 kp = 9.806 65 N
Pressure	p	pascal	Pa	m <sup>-1</sup> ·kg·s <sup>-2</sup>	N·m⁻²	1 dyne/cm <sup>2</sup> = 10 <sup>-1</sup> Pa = 10 <sup>-1</sup> N·m <sup>-2</sup> 1 atm = 101 325 Pa = 101.325 kPa 1 bar = 10 <sup>5</sup> Pa = 0.1 MPa
						1 mm Hg = 133.322 387 Pa 1 Torr = 133.322 368 Pa 1 psi = 6.894 757 kPa
Dynamic viscosity	$\eta$	pascal second	Pa·s	m <sup>-1</sup> ·kg·s <sup>-1</sup>	N·s·m⁻²	$1 P = 10^{-1} Pars = 10^{-1} Nrsm^{-2}$
Kinematic viscosity	v	square metre per second	$m^2/s$	m²⋅s <sup>-1</sup>	Pas m³kg-1 Nms kg-1	1 cP = 1 mPas 1 St = 1 cm <sup>2</sup> ·s <sup>-1</sup> = $10^{-4}$ m <sup>2</sup> ·s <sup>-1</sup>
Energy	W	joule	J	m²-kg·s-2	N·m	1 erg = 1 cm <sup>2</sup> ·g·s <sup>-2</sup> = 1 dyne·cm = $10^{-7}$ J
Power Radiant flux	P	watt	W	m²-kg·s <sup>-3</sup>	N·m·s <sup>-1</sup> J·s <sup>-1</sup>	1 cal = 4.1868 J 1 erg/s = 1 dyne·cms <sup>-1</sup> = 10 <sup>-7</sup> W = 10 <sup>-7</sup> N·m·s <sup>-1</sup> =
Absorbed dose (of radiant energy)	D	gray	Gy	m <sup>2</sup> ·s <sup>-2</sup>	J·kg <sup>-1</sup>	$10^{-7} \text{ J-s}^{-1}$ $1 \text{ rad} = 10^{-2} \text{ Gy}$
Electric potential, electromotive force	U	volt	V	m <sup>2</sup> · kg·s <sup>-3</sup> ·A <sup>-1</sup>	W·A <sup>-1</sup>	
Electric resistance	R	ohm	Ω	m²- kg·s-3-A-2	V·A <sup>-1</sup>	
Quantity of electricity	Q	coulomb	С	A·s		
Activity of a radionuclide	A	becquerel	Bq	<b>s</b> <sup>-1</sup>		1 Ci = $37 \cdot 10^9$ Bq = $37 \cdot 10^9$ s <sup>-1</sup>
Concentration (of amount of substance), molar concentration	С	mole per cubic metre	mol/m³	mol·m <sup>-3</sup>		1 mol/L = 1 M = 1 mol/dm <sup>3</sup> = 10 <sup>3</sup> mol·m <sup>-3</sup>
Mass concentration	ρ	kilogram per cubic metre	kg/m³	kg·m⁻³		$1 \text{ g/L} = 1 \text{ g/dm}^3 = 1 \text{ kgm}^{-3}$

Table 1.6.-3. – Units used with the International System

Quantity	Unit	t	Value in SI units	
	Name	Symbol		
Time	minute	min	1 min = 60 s	
	hour	h	1 h = 60 min = 3600 s	
	day	d	1 d = 24 h = 86 400 s	
Plane angle	degree	0	1° = (π/180) rad	
Volume	litre	L	$1 L = 1 dm^3 = 10^{-3} m^3$	
Mass	tonne	t	1 t = 10 <sup>3</sup> kg	
Rotational frequency	revolution per minute	r/min	$1 \text{ r/min} = (1/60) \text{ s}^{-1}$	

Table 1.6.-4. – Decimal multiples and sub-multiples of units

Factor	Prefix	Symbol	Factor	Prefix	Symbol
$10^{18}$	exa	Е	$10^{-1}$	deci	d
1015	peta	P	$10^{-2}$	centi	С
$10^{12}$	tera	Т	$10^{-3}$	milli	m
$10^{9}$	giga	G	$10^{-6}$	micro	μ
$10^{6}$	mega	M	10-9	nano	n
$10^{3}$	kilo	k	10-12	pico	p
$10^{2}$	hecto	h	10-15	femto	f
$10^{1}$	deca	da	$10^{-18}$	atto	a