01/2008:0624 corrected 6.0

## **DL-METHIONINE**

## DL-Methioninum

$$H_3C$$
  $S$   $CO_2H$  and enantiomer

C<sub>5</sub>H<sub>11</sub>NO<sub>2</sub>S [59-51-8]  $M_{\rm r}$  149.2

#### **DEFINITION**

DL-Methionine contains not less than 99.0 per cent and not more than the equivalent of 101.0 per cent of (2RS)-2-amino-4-(methylsulphanyl)butanoic acid, calculated with reference to the dried substance.

#### **CHARACTERS**

Almost white, crystalline powder or small flakes, sparingly soluble in water, very slightly soluble in alcohol. It dissolves in dilute acids and in dilute solutions of the alkali hydroxides. It melts at about  $270~^{\circ}\text{C}$  (instantaneous method).

#### **IDENTIFICATION**

First identification: A, C. Second identification: B, C, D.

- A. Examine by infrared absorption spectrophotometry (2.2.24), comparing with the spectrum obtained with *DL-methionine CRS*. Dry the substances at 105 °C.
- B. Examine the chromatograms obtained in the test for related substances. The principal spot in the chromatogram obtained with test solution (b) is similar in position, colour and size to the principal spot in the chromatogram obtained with reference solution (a).
- C. Dissolve 2.50 g in 1 M hydrochloric acid and dilute to 50.0 ml with the same acid. The angle of optical rotation (2.2.7) is  $-0.05^{\circ}$  to  $+0.05^{\circ}$ .
- D. Dissolve 0.1 g of the substance to be examined and 0.1 g of *glycine R* in 4.5 ml of *dilute sodium hydroxide solution R*. Add 1 ml of a 25 g/l solution of *sodium nitroprusside R*. Heat to 40 °C for 10 min. Allow to cool and add 2 ml of a mixture of 1 volume of *phosphoric acid R* and 9 volumes of *hydrochloric acid R*. A deep-red colour develops.

#### **TESTS**

**Solution S.** Dissolve 1.0 g in *carbon dioxide-free water R* and dilute to 50 ml with the same solvent.

**Appearance of solution.** Solution S is clear (2.2.1) and colourless (2.2.2, Method II).

**pH** (2.2.3). The pH of solution S is 5.4 to 6.1.

**Related substances**. Examine by thin-layer chromatography (2.2.27), using *silica gel G R* as the coating substance.

Test solution (a). Dissolve 0.2 g in water R and dilute to 10 ml with the same solvent.

Test solution (b). Dilute 1 ml of test solution (a) to 50 ml with water R.

*Reference solution (a).* Dissolve 20 mg of *DL-methionine CRS* in *water R* and dilute to 50 ml with the same solvent.

Reference solution (b). Dilute 1 ml of reference solution (a) to 10 ml with  $water\ R$ .

Apply separately to the plate 5  $\mu$ l of each solution. Develop over a path of 10 cm using a mixture of 20 volumes of glacial acetic acid R, 20 volumes of water R and 60 volumes of butanol R. Allow the plate to dry in air and spray with ninhydrin solution R. Heat the plate at 100 °C to 105 °C for 15 min. Any spot in the chromatogram obtained with test solution (a), apart from the principal spot, is not more intense than the spot in the chromatogram obtained with reference solution (b) (0.2 per cent).

**Chlorides.** Dissolve 0.25 g in 35 ml of *water R*. Add 5 ml of *dilute nitric acid R* and 10 ml of *silver nitrate solution R2*. Allow to stand protected from light for 5 min. Any opalescence in the solution is not more intense than that in a standard prepared at the same time in the same manner using a mixture of 10 ml of *chloride standard solution* (5 ppm Cl) R and 25 ml of *water R* (200 ppm). Examine the tubes laterally against a black background.

**Sulphates** (2.4.13). Dissolve 1.0 g in 20 ml of *distilled* water R, heating to 60 °C. Cool to 10 °C and filter. 15 ml of the solution complies with the limit test for sulphates (200 ppm).

**Heavy metals** (2.4.8). 1.0 g complies with limit test D for heavy metals (20 ppm). Prepare the standard using 2 ml of *lead standard solution* (10 ppm Pb) R.

**Loss on drying** (2.2.32). Not more than 0.5 per cent, determined on 1.000 g by drying in an oven at 105 °C.

**Sulphated ash** (2.4.14). Not more than 0.1 per cent, determined on 1.0 g.

#### ASSAY

Dissolve 0.140 g in 3 ml of *anhydrous formic acid R*. Add 30 ml of *anhydrous acetic acid R*. Immediately after dissolution, titrate with 0.1 M perchloric acid, determining the end-point potentiometrically (2.2.20).

1 ml of 0.1 M perchloric acid is equivalent to 14.92 mg of  $C_5H_{11}NO_2S$ .

#### **STORAGE**

Store protected from light.

01/2008:0560

 $M_{\star} 454.4$ 

### **METHOTREXATE**

# Methotrexatum

 $C_{20}H_{22}N_8O_5$  [59-05-2]

## DEFINITION

(2S)-2-[[4-[[(2,4-Diaminopteridin-6-yl)methyl]methylamino]benzoyl]amino]pentanedioic acid.

Content: 97.0 per cent to 102.0 per cent (anhydrous substance).

#### **CHARACTERS**

Appearance: yellow or orange, crystalline, hygroscopic powder.