A. R1 = OH, R2 = R4 = H, R3 = CO-CH₃: 3β-[(3-*O*-acetyl-2,6-dideoxy-β-D-*ribo*-hexopyranosyl-(1→4)-2,6-dideoxy-β-D-*ribo*-hexopyranosyl-(1→4)-2,6-dideoxy-β-D-*ribo*-hexopyranosyl)oxy]-12β,14-dihydroxy-5β-card-20(22)-enolide (α-acetyldigoxin),

B. R1 = OH, R2 = R3 = R4 = H: digoxin,

D. R1 = R3 = R4 = H, R2 = OH: 3β-[(2,6-dideoxy-β-D-*ribo*-hexopyranosyl-(1→4)-2,6-dideoxy-β-D-*ribo*-hexopyranosyl)oxy]-14,16β-dihydroxy-5β-card-20(22)-enolide (gitoxin),

E. R1 = R2 = R3 = R4 = H: digitoxin,

H. R1 = R2 = R3 = H, R4 = CO-CH₃: β -acetyldigitoxin,

- F. R1 = OH, R2 = H, R3 = R4 = CO-CH₃: 3β-[(3,4-*O*-diacetyl-2,6-dideoxy-β-D-*ribo*-hexopyranosyl-(1→4)-2,6-dideoxy-β-D-*ribo*-hexopyranosyl-(1→4)-2,6-dideoxy-β-D-*ribo*-hexopyranosyl)oxy]-12β,14-dihydroxy-5β-card-20(22)-enolide (diacetyldigoxin),
- G. R1 = R2 = R4 = H, R3 = CO-CH₃: 3β -[(3-*O*-acetyl-2,6-dideoxy- β -D-*ribo*-hexopyranosyl-(1 \rightarrow 4)-2,6-dideoxy- β -D-*ribo*-hexopyranosyl-(1 \rightarrow 4)-2,6-dideoxy- β -D-*ribo*-hexopyranosyl)oxy]-14-hydroxy-5 β -card-20(22)-enolide (α -acetyldigitoxin),

C. 3β,12β,14-trihydroxy-5β-card-20(22)-enolide (digoxigenin).

01/2008:0309 corrected 6.0

ACETYLSALICYLIC ACID

Acidum acetylsalicylicum

C₉H₈O₄ [50-78-2] $M_{\rm r}$ 180.2

DEFINITION

2-(Acetyloxy)benzoic acid.

Content: 99.5 per cent to 101.0 per cent (dried substance).

CHARACTERS

Appearance: white or almost white, crystalline powder or colourless crystals.

Solubility: slightly soluble in water, freely soluble in ethanol (96 per cent).

mp: about 143 °C (instantaneous method).

IDENTIFICATION

First identification: A, B.

Second identification: B, C, D.

A. Infrared absorption spectrophotometry (2.2.24).

Comparison: acetylsalicylic acid CRS.

- B. To 0.2 g add 4 ml of *dilute sodium hydroxide solution R* and boil for 3 min. Cool and add 5 ml of *dilute sulphuric acid R*. A crystalline precipitate is formed. Filter, wash the precipitate and dry at 100-105 °C. The melting point (2.2.14) is 156 °C to 161 °C.
- C. In a test tube mix 0.1 g with 0.5 g of calcium hydroxide R. Heat the mixture and expose to the fumes produced a piece of filter paper impregnated with 0.05 ml of nitrobenzaldehyde solution R. A greenish-blue or greenish-yellow colour develops on the paper. Moisten the paper with dilute hydrochloric acid R. The colour becomes blue.
- D. Dissolve with heating about 20 mg of the precipitate obtained in identification test B in 10 ml of *water R* and cool. The solution gives reaction (a) of salicylates (2.3.1).

TESTS

Appearance of solution. The solution is clear (2.2.1) and colourless (2.2.2, *Method II*).

Dissolve 1.0 g in 9 ml of ethanol (96 per cent) R.

Related substances. Liquid chromatography (2.2.29). *Prepare the solutions immediately before use*.

Test solution. Dissolve 0.10 g of the substance to be examined in *acetonitrile for chromatography R* and dilute to 10.0 ml with the same solvent.

Reference solution (a). Dissolve 50.0 mg of salicylic acid R in the mobile phase and dilute to 50.0 ml with the mobile phase. Dilute 1.0 ml of this solution to 100.0 ml with the mobile phase.

Reference solution (b). Dissolve 10.0 mg of salicylic acid R in the mobile phase and dilute to 10.0 ml with the mobile phase. To 1.0 ml of this solution add 0.2 ml of the test solution and dilute to 100.0 ml with the mobile phase.

Column:

- size: l = 0.25 m, $\emptyset = 4.6$ mm;

 stationary phase: octadecylsilyl silica gel for chromatography R (5 µm).

Mobile phase: phosphoric acid R, acetonitrile for chromatography R, water R (2:400:600 V/V/V).

Flow rate: 1 ml/min.

Detection: spectrophotometer at 237 nm.

Injection: 10 µl.

Run time: 7 times the retention time of acetylsalicylic acid. System suitability: reference solution (b):

resolution: minimum 6.0 between the 2 principal peaks.Limits:

- any impurity: for each impurity, not more than the area of the principal peak in the chromatogram obtained with reference solution (a) (0.1 per cent);
- total: not more than 2.5 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.25 per cent);
- disregard limit: 0.25 times the area of the principal peak in the chromatogram obtained with reference solution (a) (0.025 per cent).

Heavy metals (2.4.8): maximum 20 ppm.

Dissolve 1.0 g in 12 ml of *acetone R* and dilute to 20 ml with *water R*. 12 ml of this solution complies with test B. Prepare the reference solution using lead standard solution (1 ppm Pb) obtained by diluting *lead standard solution* (100 ppm Pb) R with a mixture of 6 volumes of *water R* and 9 volumes of *acetone R*.

Loss on drying (2.2.32): maximum 0.5 per cent, determined on 1.000 g by drying *in vacuo*.

Sulphated ash (2.4.14): maximum 0.1 per cent, determined on 1.0 g.

ASSAY

In a flask with a ground-glass stopper, dissolve 1.000 g in 10 ml of *ethanol (96 per cent) R*. Add 50.0 ml of *0.5 M* sodium hydroxide. Close the flask and allow to stand for 1 h. Using 0.2 ml of phenolphthalein solution R as indicator, titrate with 0.5 M hydrochloric acid. Carry out a blank titration.

1 ml of 0.5 M sodium hydroxide is equivalent to 45.04 mg of $C_9H_8O_4$.

STORAGE

In an airtight container.

IMPURITIES

A. R = H: 4-hydroxybenzoic acid,

B. R = CO₂H: 4-hydroxybenzene-1,3-dicarboxylic acid (4-hydroxyisophthalic acid),

C. salicylic acid,

- D. R = O-CO-CH₃: 2-[[2-(acetyloxy)benzoyl]oxy]benzoic acid (acetylsalicylsalicylic acid),
- E. R = OH: 2-[(2-hydroxybenzoyl)oxy]benzoic acid (salicylsalicylic acid),

F. 2-(acetyloxy)benzoic anhydride (acetylsalicylic anhydride).

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N-ACETYLTRYPTOPHAN

N-Acetyltryptophanum

 $C_{13}H_{14}N_2O_3$ [87-32-1]

 M_{r} 246.3

DEFINITION

(RS)-2-Acetylamino-3-(1H-indol-3-yl)propanoic acid.

Content: 99.0 per cent to 101.0 per cent (dried substance).

PRODUCTION

Tryptophan used for the production of *N*-acetyltryptophan complies with the test for impurity A and other related substances in the monograph on *Tryptophan* (1272).

CHARACTERS

Appearance: white or almost white, crystalline powder, or colourless crystals.

Solubility: slightly soluble in water, very soluble in ethanol (96 per cent). It dissolves in dilute solutions of alkali hydroxides.

mp: about 205 °C.

IDENTIFICATION

First identification: A. B.

Second identification: A, C, D, E.

A. Optical rotation (see Tests).

B. Infrared absorption spectrophotometry (2.2.24). *Comparison: N-acetyltryptophan CRS*.